

H₂ infrared line emission from the ionized region of planetary nebulae

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ABSTRACT

Context. The analysis and interpretation of the H₂ line emission from planetary nebulae have been done in the literature by assuming that the molecule survives only in regions where the hydrogen is neutral, as in photodissociation, neutral clumps, or shocked regions. However, there is strong observational and theoretical evidence that at least part of the H₂ emission is produced inside the ionized region of these objects.

Aims. The aim of the present work is to calculate and analyze the infrared line emission of H₂ produced inside the ionized region of planetary nebulae using a one-dimensional photoionization code.

Methods. The photoionization code Aangaba was improved in order to calculate the statistical population of the H₂ energy levels, as well as the intensity of the H₂ infrared emission lines in the physical conditions typical of planetary nebulae. A grid of models was obtained and the results then analyzed and compared with the observational data.

Results. We show that the contribution of the ionized region to the H₂ line emission can be important, particularly in the case of nebulae with high-temperature central stars. This result explains why H₂ emission is more frequently observed in bipolar planetary nebulae (Gatley's rule), since this kind of object typically has hotter stars. Collisional excitation plays an important role in populating the rovibrational levels of the electronic ground state of H₂ molecules. Radiative mechanisms are also important, particularly for the upper vibrational levels. Formation pumping can have minor effects on the line intensities produced by de-excitation from very high rotational levels, especially in dense and dusty environments. We included the effect of the H₂ molecule on the thermal equilibrium of the gas, concluding that, in the ionized region, H₂ only contributes to the thermal equilibrium in the case of a very high temperature of the central star or a high dust-to-gas ratio, mainly through collisional de-excitation.

Key words. Astrochemistry – Infrared: ISM – ISM: molecules – planetary nebulae: general – ISM: lines and bands

1. Introduction

Since the first detection of H₂ in a planetary nebula (PN) by Treffers et al. (1976), this molecule has been detected in many PNe both in ultraviolet (UV) absorption and infrared (IR) emission (e.g. Beckwith et al. 1978; Zuckerman & Gatley 1988; Aspin et al. 1993; Kastner et al. 1996; Hora et al. 1999; McCandliss et al. 2007; Sterling et al. 2005; Herald & Bianchi 2004). The H₂ molecules can be excited by UV photons, collisions with the gas, or by formation on excited levels. The contribution of each mechanism to the population of the H₂ levels depends on the physical conditions of the gas. Up to now, analyses of the excitation mechanism of the H₂ molecules producing the observed IR lines have been inconclusive (Dinerstein 1991; Shupe et al. 1998; Hora et al. 1999; Speck et al. 2003; Rosado & Arias 2003; Likkell et al. 2006; Matsuura et al. 2007).

In the literature, the H₂ IR emission lines from PNe is usually analyzed under the assumption that H₂ molecules only exist in regions where the hydrogen is neutral, such as photodissociation regions (PDRs; Tielens 1993; Natta & Hollenbach 1998; Vicini et al. 1999; Bernard-Salas & Tielens 2005), shocked regions between the expanding envelope and the wind of the pro-

genitor (Gussie & Pritchett 1988; Natta & Hollenbach 1998), or neutral clumps inside the ionized region (Beckwith et al. 1978; Gussie & Pritchett 1988; Reay et al. 1988; Tielens 1993; Schild 1995; Speck et al. 2002).

On the other hand, there is strong observational and theoretical evidence that at least part of the H₂ emission is produced inside the ionized region of PNe. Some authors have noticed that the morphology of some PNe shown by images taken in the H₂ 1-0 S(1) infrared line is very similar to those taken in [N II], [S II], and [O I] forbidden optical lines and in hydrogen recombination lines, which are produced in the ionized region (Beckwith et al. 1978, 1980; Reay et al. 1988; Webster et al. 1988; Zuckerman & Gatley 1988; Balick et al. 1991; Schild 1995; Allen et al. 1997; Guerrero et al. 2000; López et al. 2000; Arias et al. 2001; Bohigas 2001; Speck et al. 2002, 2003). Excitation temperatures of approximately 1000 to 2000 K are inferred from observations of H₂ IR emission from some PNe (Hora et al. 1999; Likkell et al. 2006, and references therein). Such high excitation temperatures indicate that the molecule is excited by a strong UV radiation field, since collisions or shocks cannot explain the presence of lines from excited levels with vibrational numbers over three (Hora et al. 1999; Black & van Dishoeck 1987). Furthermore, in a previous

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paper (Aleman & Gruenwald 2004, hereafter Paper I), we calculated the density of H₂ inside the ionized region of PNe, showing that H₂ can survive in a partially ionized region with moderate temperature where neutral and ionized species coexist. Since such regions can be large when the temperature of the ionizing star is high, the ionized region can be a potential contributor to the total H₂ line emission in some PNe.

In the present paper we study the contribution of the ionized region to the observed H₂ IR line emission of PNe. The H₂ IR line intensities are calculated with a photoionization code. The effects of the temperature and luminosity of the central star, gas density, and dust-to-gas ratio on the line emission are studied. The models are described in Sect. 2. In particular, the formalism adopted for calculating the H₂ energy level population and the intensities of the H₂ emission lines, as well as the included H₂ excitation mechanisms are described in detail. Results are discussed in Sect. 3. A summary of the conclusions and final comments are presented in Sect. 4.

2. Models

To calculate the intensity of H₂ infrared lines emitted by the studied region, the molecular density and the population of each rovibrational energy level of the electronic ground state must be known. Both the molecular density and the level population depend on the location inside the nebula, since the radiation field, and thus the physical conditions of the gas, depends on the distance to the ionizing source. The physical conditions of the nebular gas were obtained with the one-dimensional photoionization code Aangaba (Gruenwald & Viegas 1992). For a description of the code and of the calculation of the H₂ density see Paper I. For the present study, the calculation of the H₂ level population and line intensities were introduced into the numerical code. The formalism adopted for these calculations is described below.

2.1. The H₂ energy level population

In Aangaba, the population of each H₂ level can be calculated by assuming either local thermodynamical or statistical equilibrium. In the first case, the level population is set by the Boltzmann distribution and only depends on the gas temperature. We assume statistical equilibrium, where the population of the energy levels is given by the balance between the rates of population and depopulation for each level. This is equivalent to stating that there is no net variation in the population of each level, i.e.,

$$\frac{dn_w(v, J)}{dt} = 0. \quad (1)$$

In the equation above, $n_w(v, J)$ is the density of H₂ in an energy level with vibrational and rotational quantum numbers, respectively, v and J of the electronic state w . There is an equation as Eq. 1 for each level. This set of equations is linearly dependent and, to solve it, we replace one of the equations by the equation of conservation

$$\sum_{v, J} n_w(v, J) = n(\text{H}_2) \quad (2)$$

where $n(\text{H}_2)$ is the density of molecular hydrogen.

The expression on the left hand side of Eq. 1 is evaluated by the algebraic sum of the population and depopulation rates of the level (v, J) owing to the various mechanisms. To calculate

the population in the rovibrational levels of the ground state, we included several excitation and de-excitation mechanisms (radiative and collisional), as well as the possibility that H₂ is produced or destroyed at a given level by chemical reactions. Equation 1 can be rewritten as

$$\left. \frac{dn_w(v, J)}{dt} \right|_{\text{rad}} + \left. \frac{dn_w(v, J)}{dt} \right|_{\text{col}} + \left. \frac{dn_w(v, J)}{dt} \right|_{\text{chem}} = 0 \quad (3)$$

where the terms on the left hand side of the equation are the terms that correspond to level population, respectively, by radiative transitions, collisional transitions, and formation or destruction processes by chemical reactions. The processes included in the calculation are described in Sects. 2.1.1 to 2.1.3. To calculate the population of the upper electronic states, only radiative electronic transitions between the excited state and the ground electronic state are taken into account (see justification in Sect. 2.1.2).

The set of equations can be solved for the density of each level if the rate coefficient for each process and the densities of the reactant species are known. The total H₂ density, as well as the densities of H⁰, H⁺, H⁻, H₂⁺, and H₃⁺, is given by the solution of the chemical equilibrium equations (see Paper I). In the following we discuss the included mechanisms of populating and depopulating the energy levels of the H₂ molecule.

2.1.1. Radiative transitions

The population of the H₂ rovibrational levels of the electronic ground state by radiative mechanisms occurs through two main routes: electric quadrupole transitions between the rovibrational levels, involving IR photons, or electric dipole transitions to upper electronic states with posterior decay to the ground state, involving UV photons.

The transitions between two energy levels with different rotational and/or vibrational quantum numbers within the same electronic state are called rovibrational transitions. For the electronic ground state of H₂, X¹Σ_g⁺ (hereafter only referred to as X), they occur through electric quadrupole transitions, since transitions by electric and magnetic dipole are forbidden. Quadrupole transitions have much lower probabilities than the electric dipole ones. For the X state, we included 289 rovibrational levels distributed over 15 vibrational levels. The H₂ rovibrational transition probabilities were calculated by Wolniewicz et al. (1998), and the energies of the levels were kindly provided by E. Roueff (2005, private communication). The selection rules for these transitions imply that the change in the rotation quantum number must be $\Delta J = 0, \pm 2$. The lines produced by rovibrational transitions of H₂ are in the 0.28 μm to 6.2 mm range of the electromagnetic spectrum.

Transitions where the electronic state of the molecule changes, with or without change in the rotational or vibrational quantum numbers, are called electronic transitions. Electronic transitions are allowed by electric dipole and are thus very likely to occur. They are efficient in exciting and destroying H₂ molecules and must then be included in the calculation of the statistical population of level X. We included electronic transitions between the ground state and the excited electronic levels B¹Σ_g⁺ and C¹Π_u (Lyman and Werner bands, respectively). The levels B¹Σ_g⁺ and C¹Π_u are referred to hereafter just as B and C, respectively. The Λ doubling splits the energy levels of C state in two, denoted here by C⁻ and C⁺. For the B electronic level, 408 rovibrational levels, distributed over 38 vibrational levels, are taken into account; for C⁻ and C⁺, 143 and 142 rovibrational levels,

respectively, are included, distributed over 14 vibrational levels. Excitation to B and C levels can be followed by decay back to the ground state on any rovibrational level (this route is known as UV pumping) or to the vibrational continuum of X leading to the molecule dissociation. This last route is called photodissociation in two steps or Solomon process, and it is described in Sect. 2.1.3. The Einstein coefficients and the wave number of the electronic transitions have been calculated by Abgrall et al. (1994) and the photodissociation fractions by Abgrall et al. (2000). The selection rules for electronic transitions requires that the change in the rotation quantum number must be $\Delta J = \pm 1$ for Lyman band transitions, while in the case of Werner band transitions $\Delta J = 0$ for X-C⁻ transitions and $\Delta J = \pm 1$ for X-C⁺ transitions. The electronic excitation of the Lyman and Werner bands requires photons with energy between 6.7 and 15 eV.

2.1.2. Collisional transitions

Collisions may also change the H₂ energy level. In photoionized regions the average energy of the particles is less than 1.7 eV, and in regions where the H₂ density is significant the average energy is even less. Since high energies are needed for electronic transitions (> 6.7 eV), collisional electronic excitation is unlikely to occur. On the other hand, collisional rovibrational transitions may be very important in astrophysical environments and are so taken into account. We included collisions of H₂ with the main components of the gas, i.e., H, H⁺, He, H₂, and electrons. Collisions may change the total nuclear spin by inducing a temporary non-zero dipole moment in H₂ (the so called reactive collisions, in opposition to the non-reactive collisions, in which there is no change in the molecular nuclear spin).

Rate coefficients for the de-excitation of H₂ by collisions with H atoms were calculated by Martin & Mandy (1995), using the quasiclassical trajectory method, for all H₂ rovibrational levels of the ground state, for both reactive and non-reactive collisions. Le Bourlot et al. (1999) give a compilation of references for the rate coefficient of non-reactive collisional de-excitation of H₂ by H, He, and H₂ calculated by fully quantum methods, but only for H₂ vibrational levels up to $v = 3$. This method of calculation is more accurate than the quasiclassical trajectory method (D. Flower 2005, private communication) and was then preferred when available.

Rate coefficients for de-excitation of H₂ by collisions with H⁺ were calculated by Gerlich (1990) for pure rotational transitions of $v = 0$, with J_{upper} up to 9 and ΔJ up to 9. The remaining pure rotational transitions of $v = 0$ (J_{upper} from 10 to 29) were estimated by extrapolating the existing coefficients. To approximately reproduce the behavior of the available coefficients, we assume the following extrapolations:

- For $J_{lower} \leq 7$, we assume the same coefficient of the transition from the highest J_{upper} of the same parity calculated by Gerlich (1990), going to the same lower J level of the transition in question. For example, the coefficient for the transition from $J_{upper} = 11$ to $J_{lower} = 2$ is equal to the coefficient for the transition from $J_{upper} = 9$ to $J_{lower} = 2$.
- For transitions with $J_{lower} \geq 8$, we assume the same coefficient of the highest transition calculated by Gerlich (1990) involving upper and lower J levels of the same parity. For example, the coefficient for the transition from $J_{upper} = 14$ to $J_{lower} = 9$ is equal to the coefficient for the transition from $J_{upper} = 8$ to $J_{lower} = 7$.

For pure rotational transitions of levels $v \geq 1$, we assume the same coefficients as the corresponding transition of $v = 0$.

Coefficient rates for H₂ vibrational de-excitation by electrons for Δv up to 3 are given by Draine et al. (1983). Since they do not provide the J -resolved rate coefficients, we assume that the transition rate for each rovibrational transition is proportional to the Einstein coefficient of the corresponding radiative transition. Only non-reactive transitions are included ($\Delta J = 0, \pm 2$).

In the present work we assume that the coefficients for the collisional excitation of H₂ by electrons are the same as those for collisions with H⁺, because of the lack of rate coefficients for this mechanism.

2.1.3. Formation and destruction of H₂ at a given energy level

In Paper I the main processes of formation and destruction of H₂ due to chemical reactions in the ionized region of PNe were determined. Chemical reactions form or destroy H₂ in a given level. Since the H₂ molecule can be significantly produced or destroyed by these mechanisms, they must be included in the set of equations of statistical equilibrium. Unfortunately, level-resolved rate coefficients for most of the reactions are still poorly known and some assumptions must be made. The processes and our assumptions are discussed in the following paragraphs.

Photodissociation – There are two main routes for H₂ photodissociation: the direct and the two-step processes. In the direct process, the H₂ molecule is excited from the ground state to the vibrational continuum of an upper electronic state by an UV photon. The cross section for this process was calculated by Allison & Dalgarno (1969) for each of the 15 bound vibrational levels of the ground state of H₂. However, since they provide no information on the cross section for each rotational level, we assume that the coefficient rate is the same for all rotational levels of the same vibrational level.

As said above, the H₂ photodissociation in two steps begins with the excitation of the molecule from the ground electronic state to a bound rovibrational level of an upper electronic state. The subsequent decay to the vibrational continuum of the ground electronic level leads to the dissociation of the molecule. According to Stecher & Williams (1967) and Abgrall et al. (1997), this happens in approximately 11% of the excitations to the B and C levels. This process is particularly important because molecules can be dissociated by photons with energies smaller than the H ionization potential. As a result, the rate of this process is not significantly affected by the H column density, but depends mostly on the geometrical dilution of the UV radiation and on the H₂ column density (i.e., depends on the self-shielding). The Einstein coefficients for the electronic excitations and the fraction of the molecules that dissociates (dissociation fraction) have been obtained by Abgrall et al. (1994, 2000), respectively. The effect of the H₂ self-shielding on the photodissociation in two steps was included, using the formalism of Black & van Dishoeck (1987).

Photoionization – The threshold energy for the photoionization of H₂ is $E_{th}(0,0) = 15.4$ eV for the ground rovibrational level of X and is smaller for upper rovibrational levels. It is assumed here that all H₂⁺ molecular ions are formed at the ground level. For H₂ in vibrational levels $v > 4$, the threshold energy of photoionization is lower than the photoionization potential of H (13.6 eV), but since the population of such levels are often very small, particularly in the recombination zone where the

H₂ density is significant, this is not an important depopulation process. An accurate cross section for the photoionization from the vibrational ground level of H₂ is given by Yan et al. (1998, 2001)¹. We assume the same shape for the cross section for the upper vibrational levels, except that it is displaced in energy by the difference in the photoionization threshold energies of the levels. Since there are no J -resolved cross sections, it is further assumed that the photoionization rate of a given vibrational level is distributed uniformly among its rotational levels.

Formation of H₂ on grain surfaces – In Paper I, we described the model adopted for the grain surface reaction and the expression used for its rate coefficient. The coefficients for H₂ formation by this reaction for each energy level have been discussed in the literature (Takahashi & Uehara 2001; Tiné et al. 2003, and references therein), but they are not well established yet. We assumed the same level distribution of H₂ produced by this process as adopted by Black & Dalgarno (1976).

Associative detachment reaction – The associative detachment,



is a major route for the formation of H₂ in the ionized region of PNe (Black 1978, Paper I). The rate coefficients for the formation of H₂ at each bound level of the ground state by this reaction were calculated by Launay et al. (1991).

Charge exchange reactions – The H₂ molecule can be produced and destroyed by charge exchange reactions



The rate coefficients for these processes are given by Karpas et al. (1979) and Galli & Palla (1998), respectively. For the first process we assume that the level distribution of the resulting molecules has the same distribution as the one used by Black & Dalgarno (1976) for molecules formed on grain surfaces, but with the appropriate energy of formation (1.83 eV, Hollenbach & McKee 1979). For the destruction route we adopted the formula for the coefficients given by Galli & Palla (1998). Since the level-resolved coefficient rates are unknown², we assume that H₂ molecules can be destroyed by charge exchange reaction in any rovibrational level equally.

Collisional dissociation – For the collisional dissociation of H₂ by H atoms we used the formula given by Shapiro & Kang (1987) for the rate coefficient, while the coefficient is that of Donahue & Shull (1991) for the dissociation by electrons. We

assume that the H₂ molecule is dissociated following the level distribution suggested by Lepp & Shull (1983),

$$k(v, J) = k_{\text{TOTAL}} \frac{E(v, J)}{\sum_{(v, J)} E(v, J)} \quad (7)$$

where k_{TOTAL} is the total coefficient rate of collisional dissociation of H₂, and $E(v, J)$ is the energy of the level (v, J) .

2.2. H₂ and the thermal equilibrium

Thermal equilibrium is assumed for the calculation of the gas temperature, that is, the total input of energy in the gas per unit time and volume is balanced by the total loss of energy per unit volume. Besides the processes of gas heating and cooling associated with the atomic species and with the grains (for details of how dust is included in the calculations, see Gruenwald et al. 2011), in the present work, we also included the H₂ molecule contribution to the energy balance of the gas. The mechanisms of gas heating due to H₂ are collisional de-excitation, photoionization, photodissociation, formation on grain surfaces, associative detachment, and charge exchange reaction; the mechanisms of energy loss are collisional excitation, charge exchange reaction, and collisional dissociation.

2.3. H₂ infrared emission line intensities

The emissivity of a line produced by de-excitation from a given level is the product of the population of that level by the Einstein coefficient of the transition and by the energy of the emitted photon. The emissivities of the H₂ IR lines are calculated for each point of the nebula. In our models, we assume that the nebula is radiation-bounded. The intensities are calculated by integrating the emissivity over the nebular volume. It is also possible to obtain solutions for matter-bounded PNe by integrating the emissivity up to a given external radius. The gas is assumed to be optically thin for the H₂ lines.

2.4. Input parameters

A grid of theoretical models was obtained for incident radiation spectrum, gas density, and dust-to-gas typical of PNe. The central star is assumed to emit as a blackbody. The gas density and the elemental abundances are assumed constant along the nebula. We use the mean values for PNe obtained by Kingsburgh & Barlow (1994) for the abundances of He, C, N, O, Ne, S, and Ar. For Mg, Si, Cl, and Fe, whose abundances are not given by Kingsburgh & Barlow (1994), the values are those adopted by Stasińska & Tylenda (1986), in order to roughly correct for grain depletion. Since the effect of these elements is not very important for the gas cooling, their exact proportion in the form of grains is not important for our results. Amorphous carbon grains with 0.1 μm radius are assumed.

We define a standard PN with a given set of input parameters to explore their effect on the level population and on the intensity of the lines emitted by the H₂ molecule. For this, we vary one of the parameters within its typical range, while keeping the other parameters fixed. Unless otherwise noted, the parameters of the models are those of the standard PN. The input parameters for the standard PN are the following: $T_\star = 150\,000$ K, $L_\star = 3000L_\odot$, $n_H = 10^3 \text{ cm}^{-3}$, and the assumed dust-to-gas ratio is $M_d/M_g = 10^{-3}$. The temperature chosen for the central star of the standard PN is above the average for observed PNe ($\sim 80\,000$ K) in order to highlight the effects of H₂.

¹ There is a small correction to their fitting formula for the first energy interval: the coefficient of the second term should be 197.25 (H. Sadeghpour 2010, private communication).

² After the conclusion of this work we found that Savin et al. (2004) had published coefficient rates for the H₂ destruction in each vibrational level by this process. However, as we show in the following, this reaction (and its reverse) is not an important mechanism of H₂ excitation, so our conclusions will not be affected by using these new coefficients

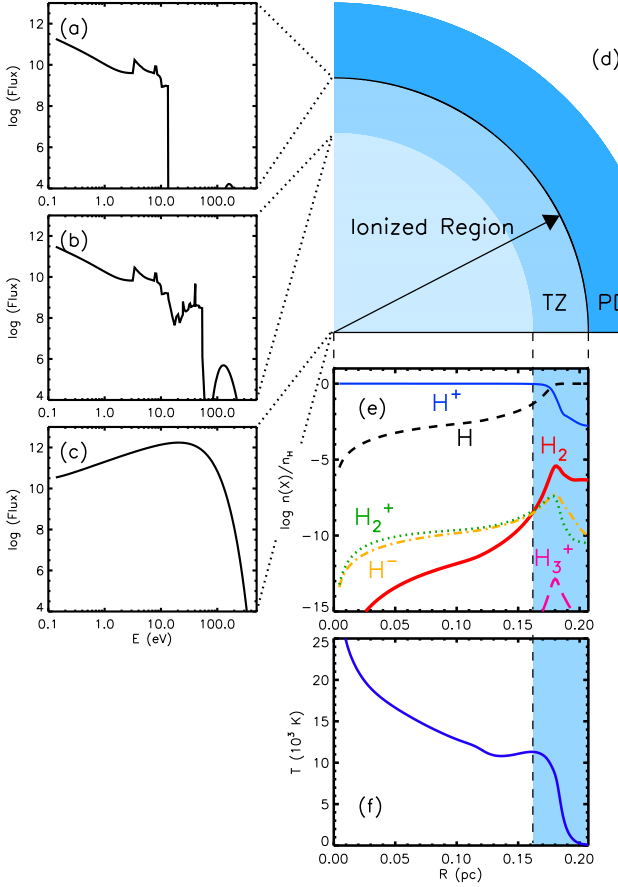


Fig. 1. The standard PN model. (a)-(c) Ionizing continuum spectra in three different positions of the nebula, as indicated in the cartoon representing the ionization structure of the standard PN in panel (d) (the PDR is not to scale). (e) Radial density profile of H⁰, H⁺, H⁻, H₂, H₂⁺, and H₃⁺ in the ionized region. Densities are relative to the total density of H nuclei. (f) Radial gas temperature profile in the ionized region. The blue band in both panels indicates the TZ. A color version of this figure is available online.

3. Results

3.1. The transition zone

In our discussion we formally define the outer boundary of the ionized region as the location where one of the following conditions is reached³: a) the fractional abundance of ionized hydrogen reaches 0.01% or b) the gas temperature is less than 100 K. The ionization degree value is chosen following Tielens (2005). Since for some PNe models this ionization degree is only reached in very low gas temperatures, we include the additional temperature stop criterion. In most of our models, the latter condition is reached first. We also define the transition zone (TZ) as the region between the location where the fraction of ionized hydrogen is 95% and the outer boundary of the ionized region. A cartoon representing the ionization structure of a PN is shown in Figure 1d.

Figure 1 shows results for the standard PN. The ionizing continuum spectrum is given for three different positions in panels

(a) to (c). Panel (c) shows the incident continuum spectrum at the inner edge of the NP. The spectrum is the geometrically diluted stellar blackbody. Panel (b) shows the continuum spectra at the inner edge of the TZ. The spectrum shows absorption mainly due to hydrogen and helium, as well as diffuse UV emission features, but there is a significant flux of UV and soft X-ray photons. These high-energy photons emitted by the central star are responsible for the formation of the TZ and are very important for the physics and chemistry in this zone. Panel (a) shows the continuum spectrum emerging from the ionized region. The flux of photons with energy above 13.6 eV is significantly absorbed, indicating the inner boundary of the PDR (see Tielens (2005)).

The radial distribution of H₂ density (relative to the total H nuclei density) for the standard PN model is shown in Fig. 1e. In the figure, R is the distance to the central star. Relative densities of H⁰, H⁺, H⁻, H₂⁺, and H₃⁺ are also included in this figure. The gas temperature profile is shown in Fig. 1f. The TZ for this model is the region indicated by the blue band in Figs. 1e and f. From these figures, it can be noticed that the relative density of H₂ molecules reaches a maximum in this warm and partially ionized region, as shown in Paper I. For typical PNe models, the maximum relative H₂ density in the ionized region is around 10⁻⁵, with the exception of models with hot central stars ($T_{\star} \geq 250\,000$ K) where the density can be orders of magnitude higher. After a slight decrease, the rising of the H₂ density towards the more neutral zone depends on the grain density (see Paper I), since the formation of H₂ on grain surfaces is the most important processes in this outer zone of the ionized region.

High-energy photons ($h\nu > 100$ eV) emitted by the central star are responsible for the formation of the TZ. They can penetrate deep into the nebulae, producing this extended region of partially ionized and warm gas. Such physical conditions favor the formation and survival of H₂, as showed in Paper I. The formation rates of H₂ molecules due to associative detachment of H and H⁻ and to charge exchange reactions between H and H₂⁺ are enhanced in a mild temperature gas where neutral and ionized species coexist. Furthermore, the destruction is not very effective because of self-shielding and shielding provided by atoms and grains present internally. Photons in the 13.6 to 100 eV range are absorbed by H and He atoms in the inner regions and do not contribute to the destruction of H₂ molecules in this region. The amount of H₂ inside the ionized region of PNe is then very sensitive to the thickness of the TZ.

The available number of high-energy photons increases strongly with T_{\star} . As a consequence, the amount of H₂ in the ionized region of PNe also increases, as can be seen in Fig. 2, where the H₂ to total H mass ratio (R_M) is shown as a function of T_{\star} . Curves for different values of L_{\star} , n_H , and M_d/M_g are plotted, respectively, in Figs. 2a, 2b, and 2c.

The size of the TZ does not depend significantly on the luminosity of the central star. However, a luminous central star produces more photons and is more effective in photoionizing and photodissociating molecular hydrogen. Consequently R_M decreases for models with higher L_{\star} (Fig. 2a).

The value of R_M also increases significantly with decreasing total H nuclei density (n_H ; see Fig. 2b). This effect is stronger in models with hotter central stars. The stronger absorption of the radiation in the gas in models with high density produces thinner TZs. It is important to notice that we are also increasing the dust density when we increase the gas density in a model, since the dust-to-gas ratio is fixed in any given model. The effect of changing the dust-to-gas ratio (M_d/M_g) on the size of the TZ is complex since dust modifies the temperature of the gas, absorbs the UV radiation, and acts as a catalyst for the molecular

³ We note that the same criteria were assumed in Paper I models, although, due to an oversight, only the ionization degree was mentioned in that paper.

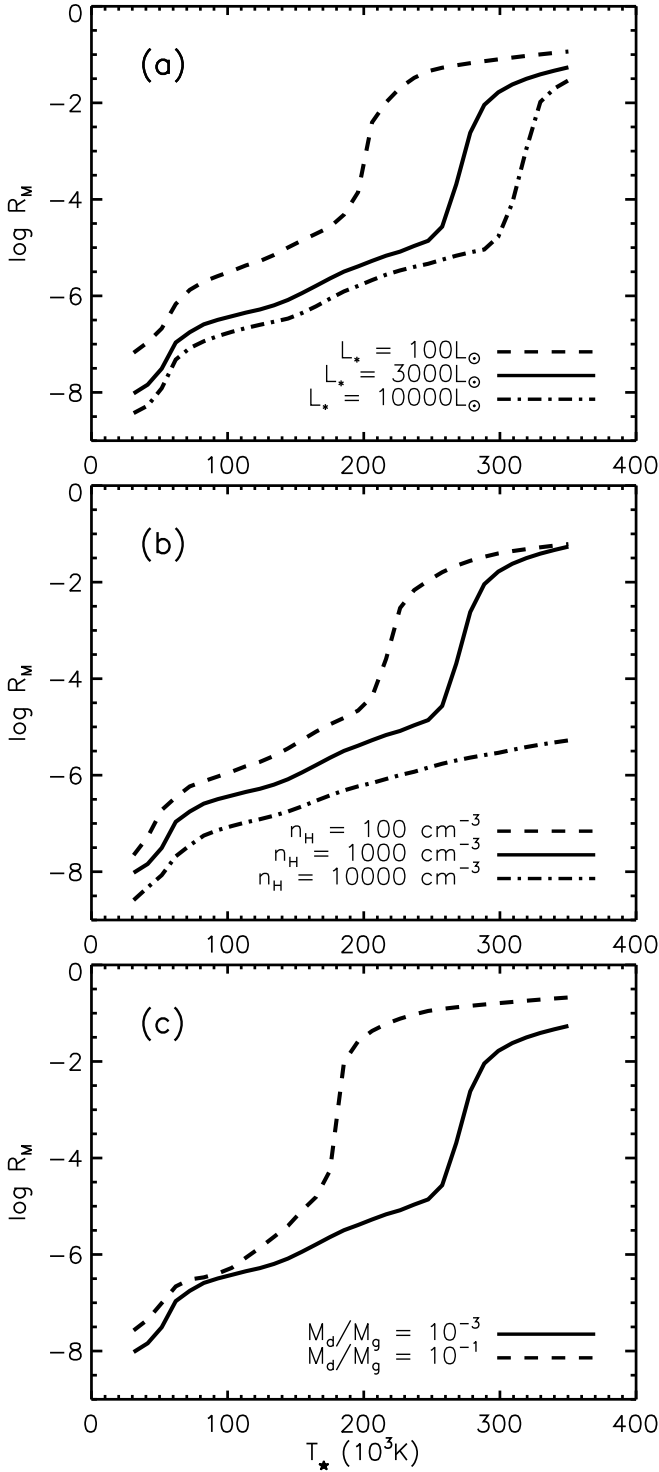


Fig. 2. H₂ to total H mass ratio (R_M) as a function of the stellar temperature. Curves for different (a) L_* , (b) n_H , and (c) M_d/M_g are shown. The standard PN values are adopted for the parameters not mentioned.

formation. As a result, the size of the TZ increases with M_d/M_g for low-temperature central stars, while the opposite effect occurs with hot stars. However, R_M increases when M_d/M_g is increased in the models, as can be seen in Fig. 2c. This is mostly due to the increase in the rate of the reaction of H₂ formation on grain surfaces. When significant density of grains is available

(i.e., $M_d/M_g > 10^{-3}$), the reaction of H₂ formation on grain surfaces is an important process of H₂ production in the outer part of the TZ.

Self-shielding is included in the present work (Sect. 2), resulting in an enhancement in R_M in comparison with the models of Paper I, particularly for models with high central star temperature, low central star luminosity, low gas density, and/or high M_d/M_g (compare Fig. 2 with their Fig. 4).

3.2. The effect of H₂ on the thermal equilibrium

Atomic species dominate the heating and cooling of the gas in the ionized region of PNe. If grains are present, they contribute significantly to the gas temperature (heating the gas) in a narrow zone near the inner border of the nebula and in the TZ. The increase in the temperature may be several thousand Kelvin in the first case and of a few hundred in the second.

The effect of H₂ on the thermal equilibrium is insignificant in most models, except in the cases of very high central star temperature and/or dust density. In these models, the net effect of the molecule is the cooling of the gas in the outer zone of the TZ, primarily through collisional excitation. For PNe with $T_* = 350\,000$ K, for example, the gas temperature is decreased by up to 5000 K, while for the standard PN the temperature structure is not altered by the presence of H₂.

3.3. The H₂ 1-0 S(1) line emission

The 1-0 S(1) line is produced by the radiative transition from the level $(v, J) = (1, 3)$ to the level $(0, 1)$ of the ground electronic state and is one of the most intense lines of H₂. Its wavelength is $2.122\,\mu\text{m}$, which is in the K atmospheric window, so it can be detected by both ground- and space-based telescopes. This line has been detected in many PNe (for example Hora et al. 1999).

The fraction of H₂ molecules at the level $(1, 3)$ as a function of the distance to central star is shown in Fig. 3 for the standard PN. This fraction varies by a factor of 5, while the H₂ density varies more than 10 orders of magnitude (Fig. 1). For all our models, the density of H₂ molecules in the level $(1, 3)$ and, as a consequence, the emissivity of the 1-0 S(1) line follow a radial profile similar to $n(\text{H}_2)$. Figure 4 shows the emissivity of the line H₂ 1-0 S(1) in the ionized region as a function of the distance to the central star for the standard PN model. The similarity of the radial profiles of the H₂ density and the 1-0 S(1) emissivity can be noticed by comparing Figs. 1 and 4. The emissivity of this line is more important in the TZ, making this zone the most important contributor to the total 1-0 S(1) line intensity of the ionized region.

According to our calculations, the level $(1, 3)$ is populated by both collisional and radiative mechanisms. The contribution of the chemical processes is negligible. For the standard PN, for example, collisions dominate the entire ionized region, while UV pumping is significant in the TZ. The relative importance of collisional over radiative processes for the level $(1, 3)$ population increases with increasing T_* and n_H , and also for decreasing L_* , although this last effect is not very significant. Collisions dominate the excitation for models with $T_* > 200\,000$ K. Formation of H₂ on grain surfaces may have some effect on the H₂ energy level distribution, particularly for the case of higher rovibrational levels and high grain-to-gas ratio, but for the level $(1, 3)$ this effect is not important. Other processes of H₂ formation and destruction of H₂ do not contribute significantly to the energy level distribution of the molecule in the ionized region.

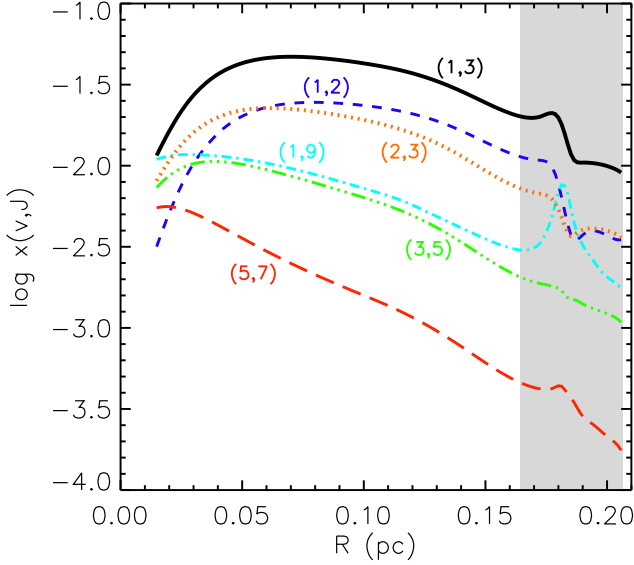


Fig. 3. Relative population of some H₂ rovibrational energy levels of the ground electronic state as a function of the distance to the central star for the standard PN model. Levels are indicated by their vibrational and rotation numbers (v, J). The gray band indicates the TZ. A color version of this figure is available online.

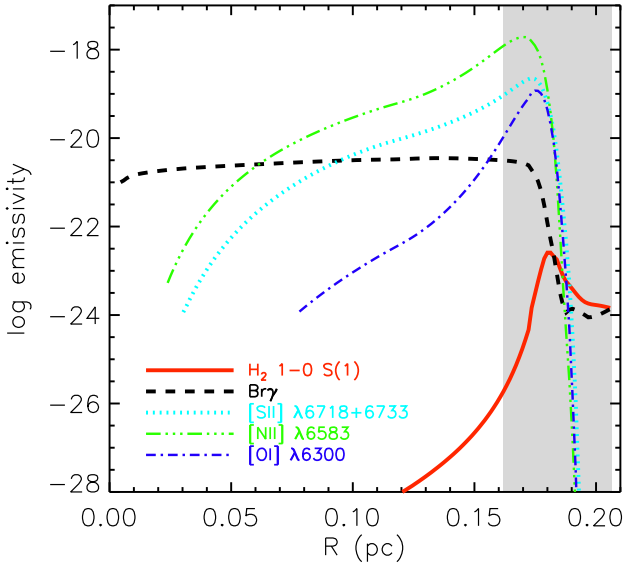


Fig. 4. Emissivities (in $\text{erg cm}^{-3} \text{s}^{-1}$) of the lines H₂ 1-0 S(1), Bry, [S II], [N II], and [O I] as a function of the distance to the central star for the standard PN model. The gray band indicates the TZ. A color version of this figure is available online.

In addition to the 1-0 S(1) line emissivity, Fig. 4 also shows the radial profiles of the emissivity of some atomic lines as calculated by our code. The lines are Bry in $2.166 \mu\text{m}$, [N II] $\lambda 6583$, [O I] $\lambda 6300$, and [S II] $\lambda\lambda 6716, 6731$. These lines are mainly produced in the region where H is ionized. According to our models, the peak in the emissivity of the 1-0 S(1) line in the ionized region is very close to the region where the atomic lines [N II], [O I], and [S II] are mainly produced.

As mentioned in the Introduction, similarities between the morphology of images in the 1-0 S(1) line and those taken in the [N II], [S II] and [O I] forbidden optical lines and hydrogen recombination lines ($H\beta$, $H\alpha$, and $\text{Br}\gamma$) have been noticed in some PNe. Since the H₂ 1-0 S(1) line can be produced in a more extended region than those lines, the similarity in the images is evidence that a neutral envelope is absent in those objects. In this case, all the H₂ emission is produced in the ionized region, and the molecule coexists with N^+ , O^0 , and H^+ in the TZ.

A correlation between the intensities of the 1-0 S(1) and [O I] lines in PNe was found by Reay et al. (1988). Although they associate this correlation with the existence of clumps inside the ionized region, it can also be naturally explained by both lines being produced in the same region, since the production of the [O I] line is very effective in the TZ, as shown in Fig. 4.

Since the 1-0 S(1) line is mostly produced in the TZ, the flux of this line is very sensitive to T_\star , as can be seen in the three top panels of Fig. 5, where the flux of the 1-0 S(1) line is shown as a function of T_\star . Curves for different L_\star (left panels), n_{H} (middle), and $M_{\text{d}}/M_{\text{g}}$ (right) are shown. The 1-0 S(1) line flux does not depend much on L_\star or $M_{\text{d}}/M_{\text{g}}$, but this flux decreases strongly if the model has a gas density higher than the standard PN value of 1000 cm^{-3} . The strong increase in the 1-0 S(1) flux with T_\star we obtain with our models agrees with the results and conclusions from Phillips (2006). After analyzing observational data from the literature, Phillips (2006) find a correlation between the detection of H₂ in PNe and their Zanstra temperatures, which he attributes to an effect of soft X-rays. The strong increase in the 1-0 S(1) flux with T_\star can also explain why the detection of H₂ is more common in bipolar PNe (Gatley's rule; Zuckerman & Gatley 1988; Kastner et al. 1996), since, according to Corradi & Schwarz (1995) and Phillips (2003), these objects typically have higher T_\star . Stanghellini et al. (2002) do not find a similar correlation in their analysis. This, however, may be due to their selection criterion that excludes PNe with high uncertainties in Zanstra temperature and, therefore, excludes high Zanstra temperature objects.

The ratio of the total luminosity in H₂ IR lines to the total luminosity of the central star is shown in the panels of the second row in Fig. 5. The sensitivity of this ratio with T_\star is evident. PNe with smaller L_\star or n_{H} also convert the stellar continuum in H₂ emission more efficiently. The change in the dust to gas ratio have a small effect in this ratio. Note that the total luminosity emitted by the H₂ lines may reach up to a few percent of the luminosity of the central star for models with a very hot and low-luminosity central star. This ratio is much lower for more typical PNe.

The 1-0 S(1) line intensity is around 1% of the total H₂ IR emission in most models, as shown in the plots of the bottom row in Fig. 5. The ratio of the 1-0 S(1) line intensity to the total H₂ IR line emission does not depend much on L_\star or $M_{\text{d}}/M_{\text{g}}$, but it increases for models with higher T_\star and decreases for models with $n_{\text{H}} > 10^3 \text{ cm}^{-3}$.

The ratio 1-0 S(1)/Bry in the ionized region as a function of the PNe model parameters is shown in Fig. 6. Models with different L_\star , n_{H} , and $M_{\text{d}}/M_{\text{g}}$ are shown in panels a, b, and c, respectively. Ratios obtained from observations are also shown. The observed values are from the following references: Beckwith et al. (1978), Isaacman (1984), Storey (1984), Webster et al. (1988), Aspin et al. (1993), Allen et al. (1997), Hora et al. (1999), Davis et al. (2003), Guerrero et al. (2000), Rudy et al. (2001), Kelly & Hrivnak (2005), and Likkell et al. (2006). In this figure we include ratios obtained from narrowband images and from long-slit spectra. In the latter

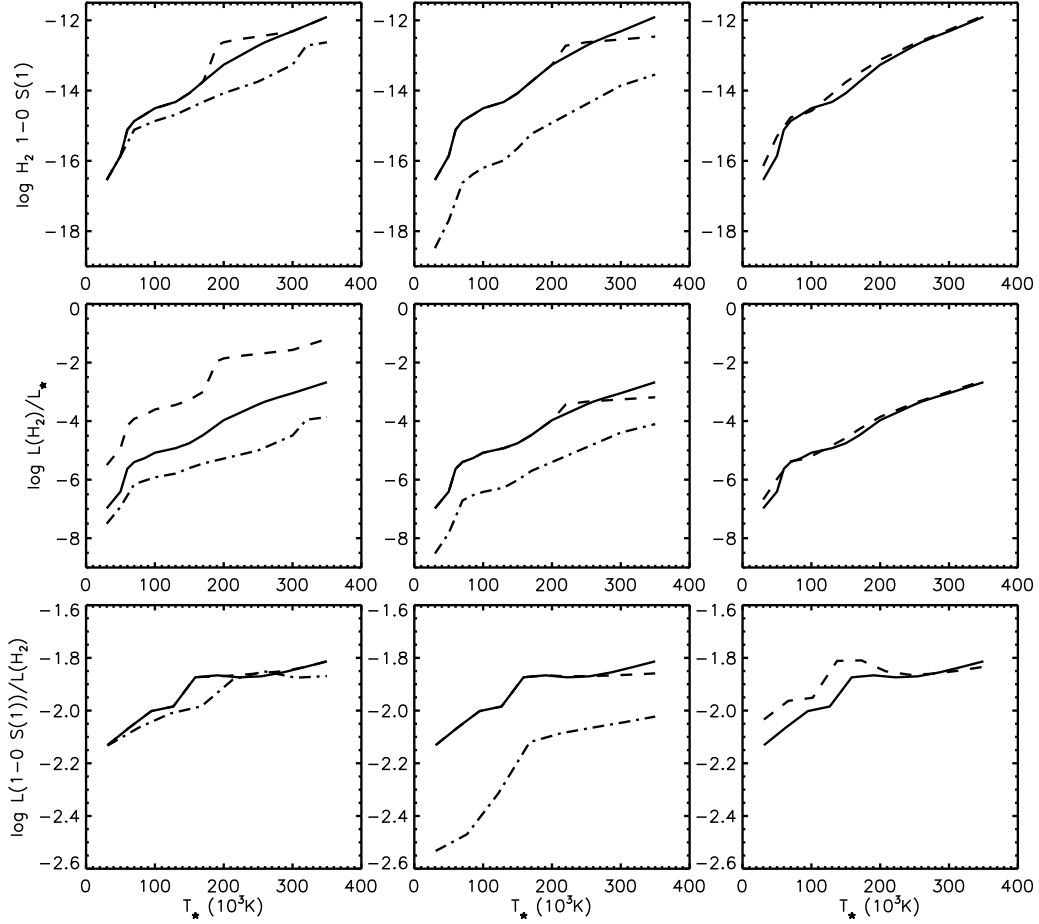


Fig. 5. Flux of the 1-0 S(1) line in arbitrary units (top row panels), ratio of the total H₂ IR lines luminosity to the central star luminosity (middle), and 1-0 S(1) luminosity to total H₂ IR lines luminosity ratio (bottom) as a function of T_\star . Curves for different L_\star , n_H , and M_d/M_g are shown, respectively, in the left, middle, and right panels in each row. The line styles are the same as in Fig. 2. In the bottom left panel, the curves for $L_\star = 100$ and $3000 L_\odot$ coincide. The standard PN values are adopted for the parameters not mentioned.

case, the slit is often positioned here the H₂ emission is more intense, so the ratio 1-0 S(1) to Br γ may be biased to higher values. In Fig. 6, T_\star represents the temperature of the stellar blackbody for models and the Zanstra or the energy-balance temperature for observations. For the Zanstra temperature, we used Tz(HeII) when available and Tz(HI) otherwise. In the few cases we did not find the Zanstra temperature, we adopted the value calculated through the energy-balance method (Preite-Martinez & Pottasch 1983). When more than one value (obtained by the same method) is found, an average is assumed. The temperatures were taken from the following references: Pottasch et al. (1978), Martin (1981), Kaler (1983), Pottasch (1984), Reay et al. (1984), Shaw & Kaler (1985, 1989), de Freitas Pacheco et al. (1986), Gathier & Pottasch (1988, 1989), Gleizes et al. (1989), Kaler & Jacoby (1989, 1991), Jacoby & Kaler (1989), Preite-Martinez et al. (1989, 1991), Walton et al. (1989), Kaler et al. (1990), Mendez et al. (1992), Stasińska et al. (1997), Bohigas (2001), and Szyszka et al. (2009).

Planetary nebulae can be radiation- or matter-bounded, depending on the amount of matter in the nebula and on the ionizing spectrum of the central star. In the first case, the ionizing spectrum cannot ionize the whole nebula. As a result, there is a neutral outer region. The emission of Br γ is produced inside

the H ionized region, while the emission of 1-0 S(1) line is produced in both the TZ and the neutral region. In our models, we assume that the nebula is radiation-bounded, therefore the line intensities are obtained by integrating the emissivity along the nebula, from the inner to the outer radius of the ionized region of the PN. Since in this work only the contribution of the ionized region is calculated, in the case of a radiation-bounded PN, only a lower limit is obtained for the H₂ line intensities and for the line ratio H₂ to Br γ . For a matter-bounded PN, the whole nebula is ionized, and our ratios are upper limits.

Figure 6 shows that the ionized region can be responsible for part or even the whole 1-0 S(1) emission of PNe. The ionized region can account for the whole emission for some PNe with high T_\star . In these cases, the hot star may be ionizing the whole (i.e., the PN is matter-bounded) or most of the nebula.

Most PNe with detected H₂ emission, however, may have a molecular envelope, which could be responsible for the differences between the calculated and observed ratios. Contamination by other nearby H₂ lines is negligible according to our models. The spectral resolution in Hora et al. (1999), from which most of the observation data were obtained, allows resolving the 1-0 S(1) line from a neighboring helium line, which may otherwise contaminate the molecular line.

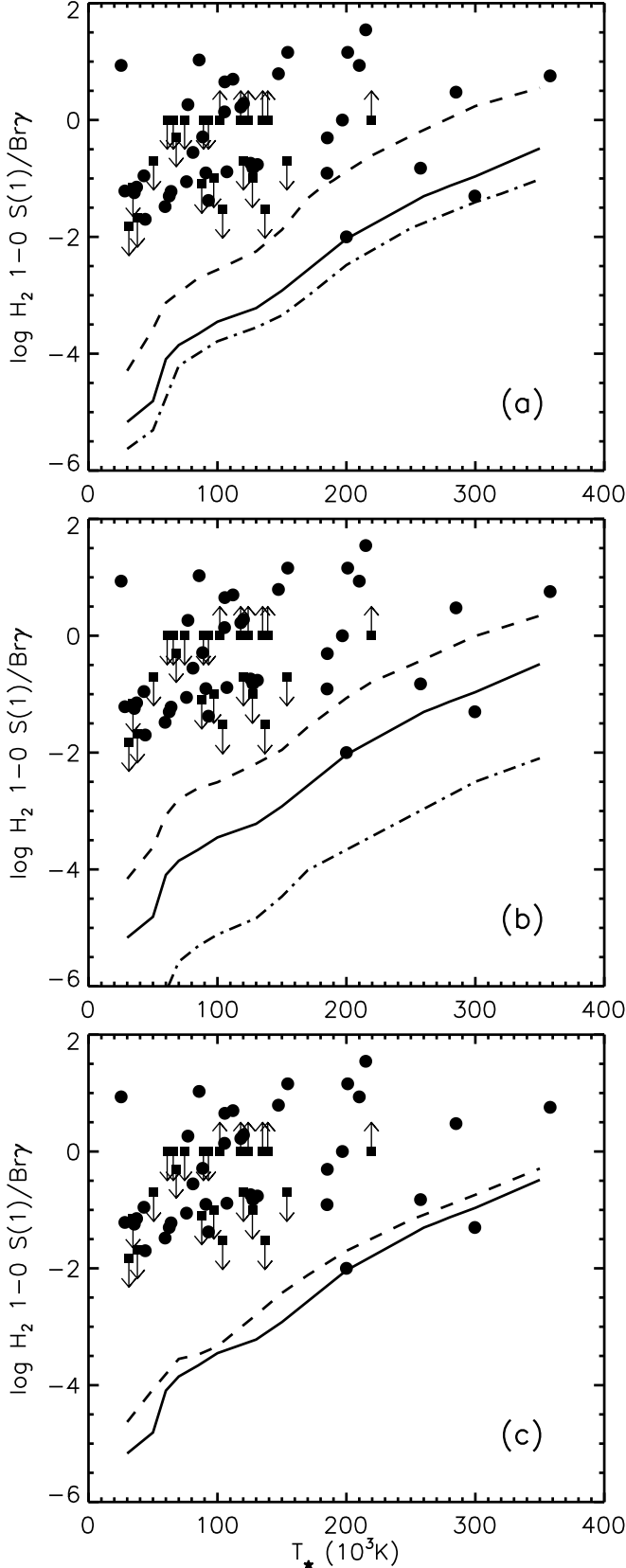


Fig. 6. Line intensity ratio of H_2 1-0 S(1) to $Br\gamma$ for the ionized region as a function of T_* . Curves for different (a) L_* , (b) n_H , and (c) M_d/M_g are shown. The line styles are the same as in Fig. 2. Observed ratios are represented by dots. Boxes with up arrows are lower limits and with down arrows are upper limits. References for the observations are given in the text. The standard PN values are adopted for the parameters not mentioned.

3.4. Other important H_2 lines

The H_2 molecule emits a few thousand rovibrational lines in the IR. According to our models, the most intense lines are emitted in the range 1 to 29 μm , most of them with wavelengths from 1 to 5 μm (near IR). Above 5 μm the most intense lines are in general pure rotational lines, particularly S(0) to S(8) of the 0-0 band. The majority of the intense lines are produced by ortho levels, especially those from higher vibrational levels, since the ortho-para ratio of H_2 is normally higher than unity. Table 1 lists those lines that are more intense than 1% of $Br\gamma$ in at least 10% of the models we ran, covering a variety of PN parameters. Assuming this criterion, the most intense lines in the ionized region are lines of the molecular bands 0-0, 1-0, 1-1, 2-0, 2-1, 3-1, 4-1, 4-2, and 5-3. Several of these intense lines have already been detected in PNe (they are indicated by an asterisk in Table 1). Several authors published observations of H_2 lines in the K band (Ramsay et al. 1993; Hora et al. 1999; Vicini et al. 1999, for example). Hora et al. (1999) report the detection of more than 50 H_2 lines, with upper vibrational levels with quantum number up to 11, in the J, H, and K bands, which are rich in H_2 lines. They are atmospheric windows, which allows both ground and space-based observations. Molecular hydrogen lines were also detected in other spectral ranges in PNe. For instance, Matsuura & Zijlstra (2005) detected the lines S(1) to S(6) of the molecular band 0-0 in observations of NGC 6302 with the Infrared Space Observatory (ISO); Bernard-Salas & Tielens (2005) published the detection of the lines 0-0 S(1) to S(7), 1-0 Q(1), 1-0 Q(3), 1-0 Q(5), 1-0 O(3), 1-0 O(4), 1-0 O(5), 1-0 O(6), and 2-1 O(3), also in observations made with ISO; the lines S(2) to S(7) of the 0-0 band were detected by Hora et al. (2006) in the Helix nebula (NGC 7293) with the Spitzer Space Telescope.

Ratios of some H_2 lines to $Br\gamma$ are shown in Fig. 7. The plots are similar to those shown in Fig. 6. As previously mentioned, in the ionized region, the production of the 1-0 S(1) line is more efficient in the TZ. Similarly, all H_2 lines are produced mostly in this zone. As a consequence, the intensity of the H_2 lines depends on the PN parameter in a similar same way as the 1-0 S(1) line, as can be seen in Fig. 7.

Ratios obtained from observations of PNe are also included in Fig. 7. References for the central star temperature are the same as in Fig. 6. The line ratios are obtained from Hora et al. (1999) and Sterling & Dinerstein (2008). Our result show that the ionized region can account for part or even all the H_2 emission of these lines, particularly for PNe with hot central stars.

3.5. Excitation mechanisms of the H_2 levels

In all the analyzed models, collisional transitions are a major mechanism for population and depopulation of the H_2 energy levels, in the whole ionized region. This mechanism dominates the excitation of the lower vibrational levels. Collisions with ionized species are important in the inner (hotter and most ionized) zone of the models. The net effect of this mechanism is to populate levels with $J < 5$ by de-excitation of the upper levels (especially with $J > 7$). Collisions with neutral species, on the other hand, are more important in the TZ, where the temperature is moderate and neutral species are abundant. The net effect of this mechanism is to de-excite the levels with lower rotational number, populating the upper rotational levels, especially those with lower vibrational numbers. The importance of collisions increase with T_* , since the TZ is more extended in this case.

The rate of H_2 rovibrational radiative excitation is negligible. On the other hand, rovibrational de-excitation following elec-

Table 1. Important H₂ lines of the ionized region of PNe

Line ^a	$\lambda(\mu\text{m})$	Line	$\lambda(\mu\text{m})$	Line	$\lambda(\mu\text{m})$
0.8286	4-1 S(7)	1.6750	1-0 S(19)	3.3718	0-0 S(23)
0.8306	4-1 S(9)	1.6877	1-0 S(9)*	3.3809	0-0 S(20)
0.8337	4-1 S(5)	1.7147	1-0 S(8)*	3.3876	0-0 S(24)
1.0519	2-0 S(10)	1.7480	1-0 S(7)*	3.4039	0-0 S(19)
1.0526	2-0 S(11)	1.7880	1-0 S(6)*	3.4379	2-1 O(5)
1.0536	2-0 S(9)	1.7904	2-1 S(9)	3.4384	0-0 S(18)
1.0576	2-0 S(8)	1.7963	2-1 S(19)	3.4856	0-0 S(17)
1.0608	2-0 S(13)	1.8358	1-0 S(5)*	3.5007	1-0 O(6)*
1.0641	2-0 S(7)	1.8528	2-1 S(7)	3.5470	0-0 S(16)
1.0733	2-0 S(6)	1.8920	1-0 S(4)	3.5926	1-1 S(21)
1.0851	2-0 S(5)	1.8947	2-1 S(6)	3.6198	1-1 S(19)
1.0998	2-0 S(4)	1.9449	2-1 S(5)*	3.6263	0-0 S(15)
1.1175	2-0 S(3)	1.9576	1-0 S(3)*	3.6522	1-1 S(18)
1.1204	3-1 S(9)	2.0041	2-1 S(4)*	3.6979	1-1 S(17)
1.1211	3-1 S(11)	2.0338	1-0 S(2)*	3.7245	0-0 S(14)
1.1240	3-1 S(8)	2.0656	3-2 S(5)*	3.7602	1-1 S(16)
1.1304	3-1 S(7)	2.0735	2-1 S(3)*	3.8075	1-0 O(7)
1.1320	3-1 S(13)	2.1218	1-0 S(1)*	3.8404	1-1 S(15)
1.1382	2-0 S(2)	2.1542	2-1 S(2)*	3.8464	0-0 S(13)
1.1397	3-1 S(6)	2.2014	3-2 S(3)*	3.8681	2-2 S(19)
1.1519	3-1 S(5)	2.2233	1-0 S(0)*	3.9391	2-2 S(17)
1.1622	2-0 S(1)	2.2477	2-1 S(1)*	3.9414	1-1 S(14)
1.1672	3-1 S(4)*	2.4066	1-0 Q(1)*	3.9968	0-0 S(12)
1.1857	3-1 S(3)*	2.4134	1-0 Q(2)*	4.0675	1-1 S(13)
1.1958	4-2 S(9)	2.4237	1-0 Q(3)*	4.0805	2-2 S(15)
1.1989	4-2 S(11)	2.4375	1-0 Q(4)	4.1622	1-0 O(8)
1.2047	4-2 S(7)	2.4548	1-0 Q(5)*	4.1810	0-0 S(11)
1.2263	4-2 S(5)*	2.4755	1-0 Q(6)	4.2236	1-1 S(12)
1.2330	3-1 S(1)*	2.5001	1-0 Q(7)	4.3138	2-2 S(13)
1.2383	2-0 Q(1)*	2.5278	1-0 Q(8)	4.4096	0-0 S(10)
1.2473	2-0 Q(3)*	2.5510	2-1 Q(1)	4.4171	1-1 S(11)
1.2616	4-2 S(3)*	2.5600	1-0 Q(9)	4.5757	1-0 O(9)
1.2636	2-0 Q(5)	2.5698	2-1 Q(3)	4.6563	1-1 S(10)
1.2873	2-0 Q(7)*	2.5954	1-0 Q(10)	4.6761	2-2 S(11)
1.2894	5-3 S(7)	2.6040	2-1 Q(5)	4.6947	0-0 S(9)
1.3107	5-3 S(5)	2.6269	1-0 O(2)	4.9533	1-1 S(9)
1.3116	4-2 S(1)*	2.6350	1-0 Q(11)	5.0529	0-0 S(8)
1.3188	2-0 Q(9)	2.6538	2-1 Q(7)	5.2390	2-2 S(9)
1.3240	3-1 Q(3)	2.6789	1-0 Q(12)	5.3304	1-1 S(8)
1.3420	3-1 Q(5)	2.7200	2-1 Q(9)	5.5115	0-0 S(7)
1.3472	5-3 S(3)	2.7269	1-0 Q(13)	5.8111	1-1 S(7)
1.3584	2-0 Q(11)	2.8025	1-0 O(3)*	6.1089	0-0 S(6)*
1.3684	3-1 Q(7)	2.8039	2-1 Q(11)	6.9091	0-0 S(5)*
1.4034	3-1 Q(9)	2.8361	1-0 Q(15)	7.2807	1-1 S(5)
1.4068	2-0 Q(13)	2.9061	2-1 Q(13)	8.0258	0-0 S(4)*
1.4295	4-2 Q(5)	2.9741	2-1 O(3)*	9.6649	0-0 S(3)*
1.4479	3-1 Q(11)	3.0039	1-0 O(4)*	12.2785	0-0 S(2)*
1.4592	4-2 Q(7)	3.2350	1-0 O(5)*	17.0346	0-0 S(1)*
1.6504	1-0 S(11)	3.3663	0-0 S(22)	28.2207	0-0 S(0)
1.6665	1-0 S(10)	3.3689	0-0 S(21)		

^a Lines with (*) are already or possibly detected in PNe. By possibly detection we mean that a blend with or contamination by other line may prevent confirmation of the detection so far.

tronic (UV pumping) or rovibrational collisional excitation is very important. The main effect of this mechanism is the excitation of H₂ from the lower rotational levels of $v = 0$ to the higher vibrational levels ($v > 3$) with $J \leq 7$. For levels with $v \geq 9$ and $J \leq 7$ UV pumping is the dominant mechanism. The contribution of the C⁻ and C⁺ states for the UV pumping rate is similar to the contribution of the B state. UV pumping is particularly important for H₂ excitation in the case of PNe with lower n_{H} and/or T_{\star} .

Formation pumping has a secondary effect on the H₂ population in the ionized region. Such an effect is only noticed for the upper levels ($v > 5$ and $J > 6$) in the outer fraction of PNe, especially for denser nebula or colder central stars. In this case, grain surface reaction and associative detachment are the main processes. Grain surface reaction may have a stronger effect for high dust density models.

The H₂ destruction processes do not affect the H₂ population in the ionized region. Photoionization has a small effect (less than 10%) in populating of the lower vibrational levels in the inner zone of the nebula. However, since the H₂ density is very low in this region, it does not affect the intensities of the H₂ lines of PNe.

4. Summary

In the present work we have studied the H₂ infrared emission from the ionized region of PNe. For this, we used the one-dimensional photoionization code Aangaba, in which we included the physics and chemistry of the H₂ molecule. This powerful tool can now be used to study the H₂ density, level population, line emission, processes of formation and destruction, and mechanisms of population and depopulation of the molecular energy levels as a function of the distance from the ionizing source of an ionized gaseous nebula.

Although there is observational evidence that at least part of the H₂ 1-0 S(1) line emission may originate inside the ionized region of PNe, the published studies of the molecular hydrogen emission of PNe do not usually take into account the contribution of the ionized region to this emission. One of the important conclusions of this work is that the H₂ emission of the ionized work can contribute significantly to the total emission observed in PNe, particularly for PNe with high T_{\star} . Comparison between the calculated and observed H₂ 1-0 S(1)/Bry ratio shows that the emission of the ionized region can be responsible for a substantial fraction of the total H₂ emission in such cases. Therefore it is important to use a code where the neutral and the ionized region are taken into account self-consistently.

In the ionized region, the H₂ IR emission lines are produced predominantly in the TZ, where the atomic lines [N II], [O I], and [S II] are significantly produced. The partially ionized and warm gas in the TZ favors the formation and survival of H₂ molecules, as well as its IR line emission. The temperature of the central star is an important factor for the H₂ density and IR line intensity, since hotter stars produce more high-energy photons than colder stars. These photons can penetrate deep into the nebulae, producing the TZ. The 1-0 S(1) line intensity increases strongly with the increase in T_{\star} in our models. This result agrees with the correlation between the detection of H₂ in PNe and their Zanstra temperatures found by Phillips (2006). Furthermore, this can explain why the detection of this H₂ line is more common in bipolar PNe (Gatley's rule), given that these objects typically have higher T_{\star} . Although Reay et al. (1988) suggest that the correlation between the intensity of the 1-0 S(1) and [O I] lines in PNe is due to the existence of clumps, such correlation can also be naturally explained by both lines being produced in the TZ.

The most intense H₂ lines are emitted in the range 1 to 29 μm , in the bands 0-0, 1-0, 1-1, 2-0, 2-1, 3-1, 4-1, 4-2, and 5-3. Several of these intense lines have already been detected in PNe. The 1-0 S(1) line is one of the more intense lines. The fraction of the 1-0 S(1) line to the total H₂ IR line emission is around 1% (within a factor of less than 10) in all our models.

Both collisions and UV pumping play important roles in the excitation of H₂ infrared lines in the ionized region. The effect

of the excitation by UV pumping is important for levels with $v > 3$. The relative importance of collisions over UV pumping increases with the increase in T_* and n_{H} . Grain surface reaction and associative detachment may only be significant for very high levels ($v > 5$ and $J > 6$), particularly in the cases of denser nebula or colder central stars. Grain surface reaction may be important for lower levels in the presence of large amounts of dust.

The effect of H₂ on the thermal equilibrium is insignificant in most models, except in the cases of a very high central star temperature or dust density, where the molecule cools the gas in the outer zone of the TZ, primarily through collisional excitation.

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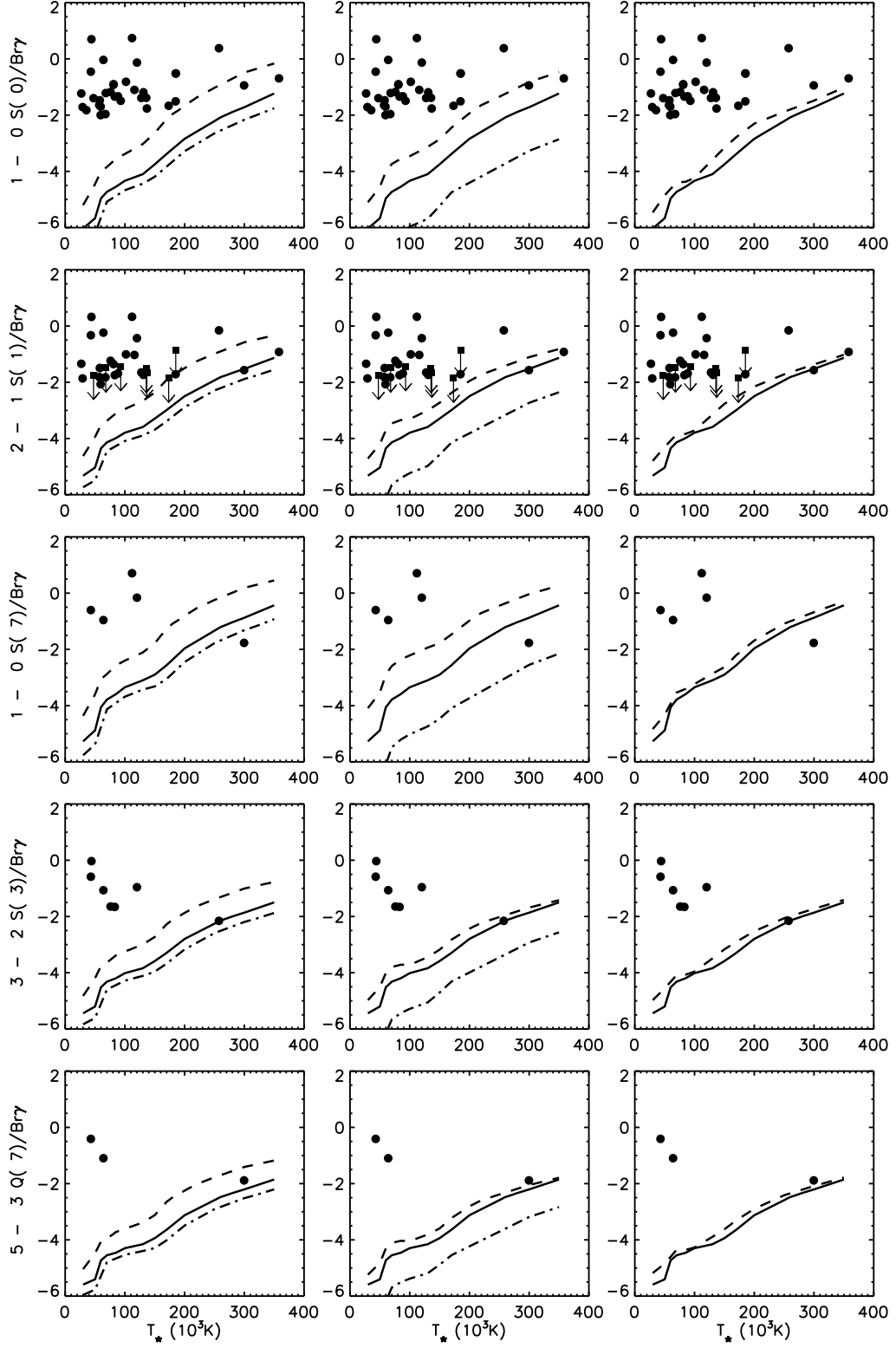


Fig. 7. Ratios of some H₂ lines to Br γ for the ionized region as a function of T_* . The models are the same as for Fig. 2. The observed ratios are represented by dots. Boxes with down arrows are upper limits. The references for the observations are given in the text.